

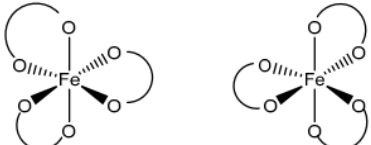
28. Chemistry of transition elements

28.4 Stereoisomerism in transition element complexes

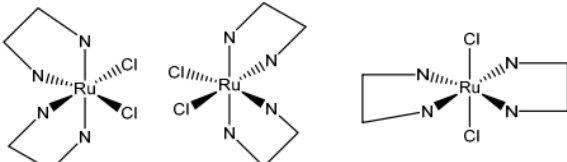
Paper 4

Marking Scheme

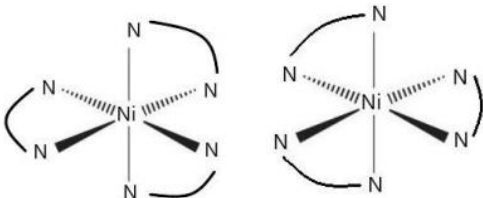
Q1.

(e)		2
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Q2.

(e)(i)		3
(e)(ii)	optical AND cis-trans / geometrical	1

Q3.

(c)		3
	octahedral with correct 3D for one $[\text{Ni}(\text{en})_3]^{2+}$ second optical isomer optical isomerism	[1] [1] [1]

Q4.

(a)	geometrical / cis-trans AND optical	1
(c)	<div style="display: flex; flex-wrap: wrap;"> <div style="width: 50%; text-align: center;"> <p>1</p> </div> <div style="width: 50%; text-align: center;"> <p>2</p> </div> <div style="width: 50%; text-align: center;"> <p>3</p> </div> <div style="width: 50%; text-align: center;"> <p>4</p> </div> </div> <p>Two correct for one mark, three correct for two marks, four correct for three marks.</p>	3

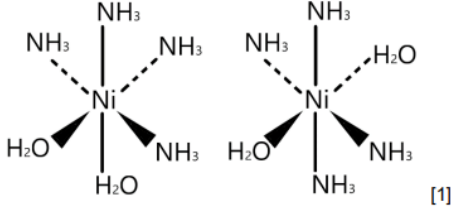
Q5.

(c)(i)	M1 (a species) that donates / uses two lone pairs M2 to form a two dative / coordinate bond to a metal atom / metal ion	2
(c)(ii)	<p>Any four from:</p> <div style="display: grid; grid-template-columns: 1fr 1fr; gap: 10px;"> <div style="text-align: center;"> <p>isomer 1</p> </div> <div style="text-align: center;"> <p>isomer 2</p> </div> <div style="text-align: center;"> <p>isomer 3</p> </div> <div style="text-align: center;"> <p>isomer 4</p> </div> <div style="text-align: center;"> <p>isomer 5</p> </div> <div style="text-align: center;"> <p>isomer 6</p> </div> </div> <p>Any two [1], any three [2], all four [3]</p>	3
(c)(iii)	oxidation state of cobalt (+)3 AND type of stereoisomerism cis-trans OR geometric(al) OR optical	1

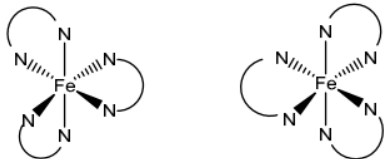
Q6.

(b)(i)	trans = non-polar cis isomer 1 = polar cis isomer 2 = polar ALL [1]	1
(b)(ii)	optical isomers / non-superimposable mirror images [1] ALLOW enantiomers / they rotate polarised light differently	1


Q7.

(b)(v)	$[\text{Rh}(\text{en})_2\text{Cl}_2]^+$ AND optical AND cis / trans OR geometric [1]	1
(b)(vi)	 <p>cis/trans OR geometric [1]</p>	2

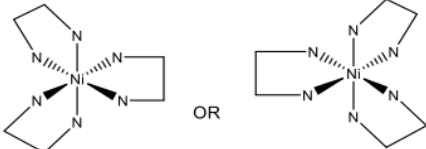
Q8.

(b)(ii)	 <p>M1 one correct 3D diagram with three bipy ligands M2 both diagrams correct M3 optical isomerism</p>	3
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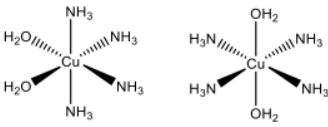
Q9.

(e)	 <p>M1 one correct 3D diagram with the six correct ligands M2 both 3D diagrams correct</p>	2
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Q10.

(b)(ii)		1
(b)(iii)	octahedral AND optical isomerism	1

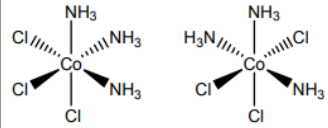
Q11.

(b)(i)	 <p>cis and trans drawn</p>	1
(b)(ii)	<i>cis</i> identified as polar AND <i>trans</i> identified as non-polar conditional on diagram	1

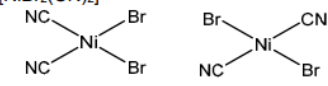
Q12.

(c)(i)	forms 2 dative / coordinate bonds	1
(c)(ii)	one isomer correct with correct 3D [1] three isomers different with correct 3D [1]	2
(c)(iii)	e.g. 1 and 2 optical [1] e.g. 1 or 2 and 3 cis-trans / geometric [1]	2


Q13.

(c)	 <p>geometric ALLOW cis-trans Two correct responses = 1 mark Three correct responses = 2 marks</p>	2
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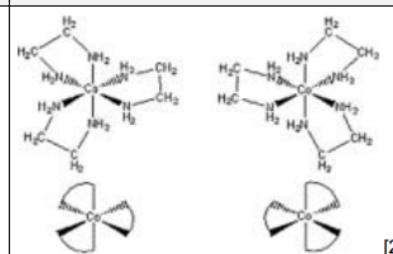
Q14.

(d)	<p>M1 two correct structures cis & trans for $[\text{NiBr}_2(\text{CN})_2]^{2-}$</p>  <p>M2 type of stereoisomerism: cis-trans/geometric</p>	2
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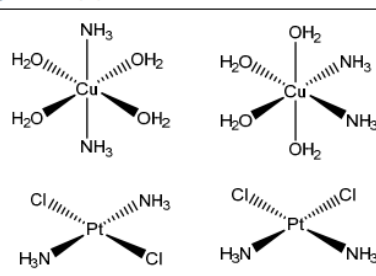
Q15.

(c)(i)	 <p>both correct [1]</p>	1
(c)(ii)	square planar [1]	1

Q16.

(d)(ii)	 <p>[2]</p>	2
(d)(iii)	optical [1]	1

Q17.

(e)(i)	cis-trans isomerism [1] ALLOW geometric(al)	1
(e)(ii)	 <p>one correct pair [1] two correct pairs [2]</p>	2